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Problem #5: Given That The Molar Mass Of  $\text{Na}_2\text{SO}_4 \cdot \text{NH}_2\text{O}$  Is 322.1 G/mol, Calculate The Value Of N.  
Solution: 1) The Molar Mass Of Anhydrous  $\text{Na}_2\text{SO}_4$  Is: 142.041 G/mol 2) The Mass Of Water In One Mole Of The Jun 2th, 2022

## **CHEM 10100 - Inquiries Into The Nature Of Matter**

In The First Day Of Lab You Will Be Require To Have: •  
Chemistry 101 “Inquiries into The Nature Ofmatter”  
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### **Naming Acids And Hydrates Worksheet**

Naming Bases Most Strong Bases Contain Hydroxide, A Polyatomic Ion. Therefore, Strong Bases Are Named Following The Rules For Naming Ionic Compounds. For Example, NaOH Is Sodium Hydroxide, KOH Is Potassium Hydroxide, And Ca(OH)<sub>2</sub> Is Calcium Hydroxide. Weak Bases Made Of Ionic Compounds Are A Feb 5th, 2022

### **Hydrates 2 - 11HZ È(òFp**